

A Study of Combating Climate Change with Carbon Capture and Sequestration

Kapil Nichani, Shalmali Bapat, Dr. Anita Kumari

Abstract— Carbon Capture has been framed 25 years ago as a theoretical concept, but CCS as a green- house gas abatement method is a relatively new development. The International Energy Agency and the Carbon Sequestration Leadership Forum recognizes the role of CCS to bring about equilibrium in carbon cycle. It aims to bring about public acceptance for CCS. Although technology has been developed extensively for carbon capture much work is required to address numerous issues like – Monitoring of CO₂ behaviour underground and the leak potentials over ages. Engineering and economic factors also need to be restructured to find a remedy to high costs of the current technologies involved. A major hurdle is retrofitting the current power plants with CCS capabilities. Compared to coal fired power plants, gasification process enables easier and less costly capture of CO₂. The Petra Nova project which will be the world's largest carbon capture facility is being currently built. Such a project will serve as a benchmark to gain widespread acceptance and awareness for this technology. Carbon Capture and Sequestration does not serve as a silver bullet but it is a significant method to combat climate change. CCS along with developments in cleaner energy sources will enable the world to lead towards a balanced carbon system.

Index Terms— Green-house gas, Enhanced Oil Recovery, Carbon Capture, Sequestration, Climate change

I. INTRODUCTION

Global Warming has been the most contentious issue of the 21st century caused by anthropogenic Green House Gases. Reduction of such gases has been the aim of various research groups. Efforts have been made to develop alternatives to fossil fuels. Large number of green house gas emitting facilities will continue to exist for times to come. Political agreements reached through Global co-operation have been sought after the Kyoto Protocol which targets reduction of green house gases; carbon dioxide being one of them. Scientists claim water vapour is the most abundant green house gas, as it amplifies the effect of carbon dioxide, making carbon dioxide lethal to environment and ecology. According to recent data, 32% of the total CO₂ production is from power plants. China emits more CO₂ than the US and Canada put together. India is world's third biggest emitter of CO₂ pushing Russia to fourth [1].

Manuscript received September 22, 2014.

Kapil Nichani, Department of Chemical Engineering, Thadomal Shahani Engineering College, Bandra, Mumbai, India. (Mob: +91 7738788556).

Shalmali Bapat, Department of Chemical Engineering, Thadomal Shahani Engineering College, Bandra, Mumbai, India. (Mob: +91 9867118555).

Dr. Anita Kumari, Associate Professor, Department of Chemical Engineering, Thadomal Shahani Engineering College, Bandra, Mumbai, India.

This unparalleled rise of CO₂ concentration in the atmosphere calls for the need to address the issue. Carbon Capture and Storage (CCS) is a series of technologies for CO₂ control that promises trapping up to 90% of the CO₂ emissions from power stations and industrial sites.

It collectively involves accumulating, transporting and burying the CO₂ such that it does not escape into the atmosphere and subsequently contribute to climate change. However, there much deliberation and research on the long-term consequences of CO₂ trapped underground is necessary. The high costs involved with CCS have proved to be a deterrent in its implementation. Researchers and policy makers are thus urged to make the technology cost effective considering its significant role in combating climate change.

II. CARBON AND CLIMATE CHANGE

The Earth witnessed the last ice age with atmospheric CO₂ levels of being around 280 ppmv (parts per million by volume). CO₂ concentrations have been on a rise since then. The recent rise is shown in **Fig. 1**, collected by Keeling (1960) since 1958 (shown in blue) and expanded on the right-handgraph. Ensuing to rise in CO₂ levels, there has been a substantial increase in the mean global temperature. The nexus between elevated levels of CO₂ in the Earth's atmosphere and rising temperatures is well established. Earth radiates much of its heat energy in form of terrestrial radiation in the infrared region.

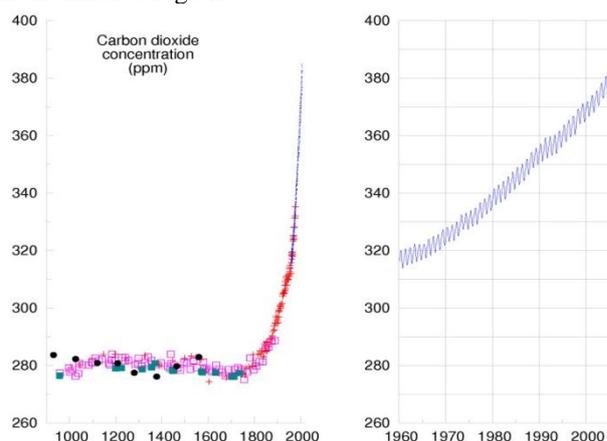


Figure 1¹: Carbon dioxide Concentration rise

III. SOIL CARBON SEQUESTRATION

Excretions and remains of plants and animals add carbon to soil on a continual basis. CO₂ is released from soil on an on-going basis, as microbes break down soil organic matter (SOM). Nature has a well an established carbon cycle.

¹Keeling and Whorf (2005); Neftel et al (1994); Etheridge et al (1998); Siegenthaler et al (2005); Indermuhie et al (1999)

Equilibrium in soil is reached when inputs equal outputs. Ecosystems like forests control Earth's climate by adding and removing CO₂ from the atmosphere. These forests and grasslands which primarily form the earth's terrestrial ecosystems store much more carbon than what the atmosphere does [2]. CO₂ is thus kept out of the atmosphere, where it would contribute to climate change. Numerous plants and soil over time store carbon, thereby sequestering additional carbon. Thus aping the technique of nature, the high tech Carbon Capture and Storage aims to bring equilibrium in the Carbon cycle.

IV. CARBON DIOXIDE CAPTURE AND SEQUESTRATION

Humanity is slowly moving into a Carbon constrained world. The technique of Carbon Capture and Sequestration (CCS) will help us to sustain the benefits of using hydrocarbons to generate energy. The use of fossil fuels will determine the future levels of atmospheric CO₂, and consequently the future levels of the associated global warming. Fossil fuels serve as a dominant and relatively inexpensive source of energy and will continue to be widely used for remainder of the century.

CO₂ is inevitably produced as long as fossil fuels are used; if we are to prevent this CO₂ reaching the atmosphere, Carbon Capture and Sequestration (CCS) will be absolutely essential. In a *Special Report on Carbon Dioxide Capture and Storage*, the International Panel on Climate Change (IPCC) recognized CCS as one of the tools that can be used to reduce GHG releases to the environment [3]. CCS can be considered a 'waste management strategy' for carbon dioxide. It does not reduce the production of CO₂, but it provides a reservoir to prevent it from destroying the environment.

CCS is a three-step process that includes:

- CO₂ capture from power plants or industrial production houses
- Transportation of the captured and compressed CO₂ (usually through pipelines).
- Injection of the gas and geologic sequestration (also referred to as storage) of the CO₂ into deep underground rock formations.

Injection is done 7000 feet into the earth below the impermeable, non-porous layers of rock that trap the CO₂ and prevent it from transferring upward.

Numerous commercial operations are currently storing CO₂ around the world.

- EnCana's project at Weyburn in Saskatchewan uses CO₂ gas to enhance the recovery of oil. CO₂ derived from a gasification plant in North Dakota is injected into oil reservoirs to push it to the surface.
- In Salah saline formation project in Algeria [4].
- Sleipner project in the Norwegian North Sea (**Fig. 2**) of Statoil has successfully pumped more than 1 million tons of CO₂ per year for the last 10 years into the deep saline Utsira formation, which lies approximately 1000 m beneath the seabed and is capped by a low-permeability shale layer. CO₂ is pumped deep into saline formations and eventually the gas moves up until it reaches the impermeable shale level.

These locations are an international focus for research into how CO₂ can be stored and monitored underground.

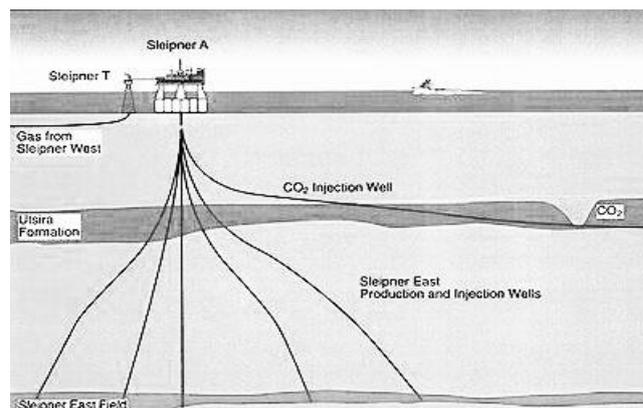


Figure 2: Schematic of Statoil's Sleipner CO₂ Injection Project
(Diagram courtesy SINTEF)

CCS when applied to a 500 MW coal-fired power plant, which emits roughly 3 million tons of CO₂ per year, [5] the amount of GHG emissions avoided (with a 90% reduction efficiency) is predicted to be equivalent to [6]:

- Planting more than 62 million trees, and waiting at least 10 years for them to grow.
- Avoiding annual electricity-related emissions from more than 300,000 homes.

V. CARBON DIOXIDE TRAPPING

A. Post Combustion

CO₂ is separated from the flue gas of the power station by bubbling the gas through an absorber column packed with liquid solvents (monoethanolamine) that preferentially take out the CO₂. The absorber column is regenerated by passing steam at around 150°. This releases the trapped CO₂, which can then be transported for storage elsewhere. Such an extraction process is inefficient and considerable energy is utilized for steam regeneration.

B. Oxyfuel

Burning of fossil fuels in normal air produces a mixture of flue gases. More energy is required to separate CO₂ from such a mixture. An alternative method is to burn the fossil fuel in an atmosphere of pure oxygen. This way all waste gases will comprise of CO₂ and water vapour. Water vapour can be condensed and CO₂ can be piped or transported directly to a storage facility. In the oxyfuel system, the air fed into the boiler has to be separated into liquid oxygen, gaseous nitrogen, argon and other trace gases. There is still a considerable energy cost associated with the cryogenic separation to obtain pure oxygen atmosphere.

C. Pre Combustion

Coal-gasification combined cycle power plants involve gasification of coal to produce a synthetic gas made from carbon monoxide and hydrogen. The former is oxidized with water to produce CO₂, which is captured. The hydrogen is usually diverted to a turbine where it can be burned to produce electricity. This hydrogen can also be bled off and used as a fuel. One disadvantage of the pre-combustion method is that it cannot be retro-fitted to the conventional coal power plants.

Such an extraction of the CO₂ from the pre-combustion stream is potentially more efficient than in the post-combustion case. The overall cost of capture would be reflected in an increase of the retail price of the electricity produced of about 20% [7].

Since these processes separate CO₂ with other components in the emissions and reduces the entropy, a certain amount of energy is required. The minimum energy required by the second law of thermodynamics is [8]

$$W_{min} = RT' \left[X_{CO_2} \ln(X_{CO_2}) + X_{gas} \ln(X_{gas}) \right]$$

Where

X_{CO₂} = Mole fraction of CO₂

X_{gas} = Mole fraction of other gases without CO₂

Thus it can be concluded that less energy is required for higher concentration of CO₂ in the emissions.

Considering the exorbitantly high energy consumption for separating CO₂ from air, CCS technologies evidently are only feasible when the CO₂ concentration in air is rackingly high. Energy required to separate CO₂ from emissions can be derived from the boilers in the power plants.

VI. CARBON DIOXIDE TRANSPORTATION

CO₂ can be transported by pipelines or ships. The density of CO₂ should be that of a liquid during transport [9] for maximum efficiency. Pipelines are usually very long and the transport temperature of CO₂ has to be close to ambient soil temperature, ranging from a few degrees below zero to 20 °C [10]. If the pressure of CO₂ is maintained over the critical pressure of 7.38 MPa, there will be no gaseous CO₂ in the pipeline from temperature fluctuations. Since CO₂ is corrosive, the risk of leaks may be greater than with other substances transported by pipeline. The gas must be dried to reduce the risk of corrosion. Local geological conditions play a crucial role to determine the piping costs.

For shipping, CO₂ is kept quite cold at reduced pressures. The temperature in that case is generally around -54 to -50 °C and the pressure is around 0.6 to 0.7 MPa [11].

VII. STORAGE OF CARBON DIOXIDE

Two locations have been found appropriate for storing CO₂ – oceans and in geological structures beneath the Earth's surface. Sequestration in the oceans is troubled with political and technical problems. Geological sequestration involves pumping of supercritical CO₂ deep into aquifers. Spent oil and gas reservoirs are usually preferred as they are more docile. Spent hydrocarbon reservoirs provide capacity for hundreds of billions of tons of CO₂, while deep saline aquifers provide capacity for a hundred times this amount [11]. Due to the abundance in availability, CO₂ storage in saline aquifers is the most studied storage form.

The sedimentary rocks serve as the most suitable kind of CO₂ storage for spent oil and gas reservoirs. Their high

porosity and permeability assist in the storage. This way CCS may thus also help to recover natural gas and oil reservoirs.

Evidence from modeling studies and observation of natural CO₂ reservoir suggest that the gas can be confined in geological reservoirs for time scales well in excess of 1000 years and that the risks of leakage from geological storage can be small [12].

VIII. ENHANCED OIL RECOVERY (EOR)

CO₂ is injected into already developed oil fields where it mixes with and releases the oil from the formation, thereby freeing it to move to production wells. CO₂ that emerges with the oil is separated in above-ground facilities and re-injected into the formation. CO₂-EOR projects are essentially a closed-loop system where the CO₂ is injected, produces oil, is stored in the formation, or is recycled back into the injection well.

The gas is primarily used as an injectant, which changes the properties of crude oil and allow it to flow more freely within the reservoir. The gas detaches the oil from the rock surfaces, and causes it to flow more freely within the reservoir to producer wells. CO₂-EOR typically produces between 4-15 percent of the original oil in place. [13].

CO₂-EOR provides a laboratory for observation, testing, scaling and modeling of technologies required for routine sequestration that is driven by the commercial incentive to improve recovery of the oil from older reservoirs. Moreover, in a world where CO₂ is much more readily and cheaply available, there will be an encouragement to use CO₂ earlier in the oil-recovery stage to better exploit diminishing resources.

Enhanced Oil Recovery provides for economic incentive to implement CCS technologies with the added benefits of greater yield of oil extraction.

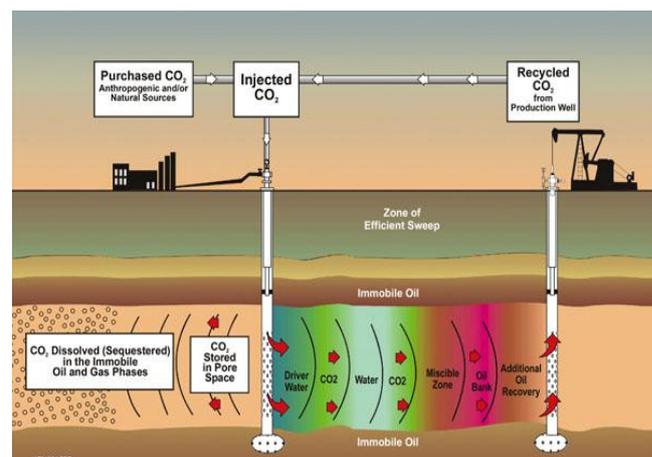


Figure 3²: Enhanced Oil Recovery

IX. PETRA NOVA CARBON CAPTURE PROJECT

A commercial scale post-combustion carbon capture and sequestration (CCS) project is being developed at NRG Energy's WA Parish coal-fired Generating Station. It is one of

²Source: Advanced Resources International and Melzer Consulting, Optimization of CO₂, Storage in CO₂, EOR Projects, Prepared by UK Dept. of Energy and Climate Change, November 2010.

the ten biggest power stations in the US. The plant is located in Thompsons, Fort Bend County. The WA Parish CCS project is also known as the Petra Nova Carbon Capture Project. It is estimated that it will capture 90% of the carbon dioxide (CO₂) from a 240MW stream of flue gas from the power station's existing 610MW coal-fired unit. Approximately 1.6 million tons of CO₂ will be pumped each year. Carbon dioxide will be purified and compressed into a liquid to be piped 80 miles away to the West Ranch Oil Field for enhanced oil recovery (EOR). The Petra Nova Project will eventually be the world's largest carbon capture facility [14].

The Petra Nova system will collect carbon dioxide using gas absorption with help of amines. Petra Nova captures carbon dioxide to help people mine ever more oil, which creates its own emissions once it's burned to make electricity. EOR will be able to recover an estimated 60 billion barrels of oil from the field. The Petra Nova CCS project is expected to boost the field's production from 500 barrels per day (bpd) to 15,000 bpd.

The project's carbon capture capacity is expected to be 4,776 tonnes per day [15]. A cogeneration plant consisting of a combustion turbine and a heat recovery boiler will provide the thermal energy required for the CO₂ capture and compression system's operation.

X. LEAKAGE RISKS

There exist potential risks in pipeline transportation of carbon dioxide gas as well as in geological storage.

Cracks and crevices in geological storage volume can leak the gas into upper formations. The subsequent migration of the released CO₂ will then be determined by the permeability of these overlying formations. Further injection of CO₂ may increase the failure intensity leading to migration of CO₂ with the flow of brine to locations where it can escape to the atmosphere or to a body of water. CO₂ entering water bodies can have severe environmental consequences, as the gas makes water acidic. Furthermore it may also dissolve hazardous constituents such as Arsenic and lead from minerals into the water bodies. The injection sites are thus chosen in which the geological formations provide several seals, consisting of shale with very low permeability, over the reservoir [16]. The risks potentially increase many fold if CO₂ is co-sequestered with other hazardous gas such as H₂S and SO₂, since they are more poisonous and caustic in solution.

A critical aspect of CO₂ storage is thus the monitoring of these possible processes.

XI. CONCLUSIONS

Carbon Capture and Sequestration is a promising technology to combat climate change. Coal Gasification and Pre-combustion techniques will prove to be a boon for CCS. With implementation of projects like Petra Nova, widespread recognition and awareness is required. Policy makers and researchers need to work in synergy to promote CCS. Critics point at the high risks and high costs pertaining CCS, but man will otherwise pay a greater price in form of ecological collapse if balance is not restored.

REFERENCES

- [1] "World carbon dioxide emissions data by country: China speeds ahead of the rest," [Online]. Available: <http://www.theguardian.com/news/datablog/2011/jan/31/world-carbon-dioxide-emissions-country-data-co2>. [Accessed 15 March 2014].
- [2] L. Murari, H. Hideo and T. Kiyoshi, "Future climate change and its impacts over small," vol. 19, p. 179–192, 2002.
- [3] I. P. C. C. (IPCC), "Special Report on Carbon Dioxide Capture and Storage: Summary for Policymakers," 2005.
- [4] F. A. Riddiford, I. Wright, C. D. Bishop and A. A. Espie, "Monitoring geological storage: the in salah gas CO₂ storage project," in *7th Int. Conf. on Greenhouse Gas Control Technologies*, 2005.
- [5] Massachusetts Institute of Technology, "The Future of Coal: Options for a Carbon-Constrained World," 2007.
- [6] U. Environmental Protection Agency, "Greenhouse Gas Equivalencies Calculator," [Online]. Available: <http://www.epa.gov/cleanenergy/energy-resources/calculator.html#results>. [Accessed 15 September 2014].
- [7] R. Socolow, "Can We Bury Global Warming?," *Scientific American*, July 2005.
- [8] G. A. Meehl, "Climate Change 2007: The Physical Science Basis," Intergovernmental Panel on Climate Change, 2007.
- [9] Z. X. Zhang, G. X. Wang, P. Massarotto and V. Rudolph, "Optimization of pipeline transport for CO₂ sequestration," *Energy Conversion and Management*, vol. 47, no. 6, pp. 702-715, 2006.
- [10] S. McCoy and E. S. Rubin, "An Engineering-Economic Model of Pipeline Transport of CO₂ With Application to Carbon Capture and Storage," *International Journal of Greenhouse Gas Control*, vol. 2, 2008.
- [11] B. Metz, "IPCC Special Report on Carbon Dioxide Capture and Storage," Cambridge University Press, 2005.
- [12] E. J. Wilson, T. L. Johnson and D. W. Keith, "Regulating the Ultimate Sink: Managing the Risks of Geologic CO₂ Storage," *Environmental Science and Technology*, vol. 37, no. 16, p. 3476, 2003.
- [13] A. R. I. (ARI), "Improving Domestic Energy Security and Lowering CO₂ Emissions with "Next Generation" CO₂-Enhanced Oil Recovery," 2011.
- [14] F. Diep, "Enormous Carbon Capture Project Begins Construction In Texas," 9 September 2014. [Online]. Available: <http://www.popsci.com/article/science/enormous-carbon-capture-project-begins-construction-texas>. [Accessed 10 September 2014].
- [15] power-technology.com, "WA Parish Carbon Capture Project, Texas, United States of America," [Online]. Available: <http://www.power-technology.com/projects/wa-parish-carbon-capture-project-texas/>. [Accessed 20 March 2014].
- [16] S. Bachu, "Sequestration of CO₂ in Geological Media: Criteria and Approach for Site Selection in Response to Climate Change," *Energy Conversion and Management*, vol. 41, 2000.

Author Details:

Kapil Nichani, Department of Chemical Engineering, Thadomal Shahani Engineering College, Bandra, Mumbai, India. (Mob: +91 7738788556. Final year Chemical Engineering Student. The above manuscript is a part of research project whose further extension is to study and conduct experiments on CO₂ absorption which is currently undergoing.

Shalmali Bapat, Department of Chemical Engineering, Thadomal Shahani Engineering College, Bandra, Mumbai, India. (Mob: +91 9867118555, Final year Chemical Engineering Student. The above manuscript is a part of research project whose further extension is to study and conduct experiments on CO₂ absorption which is currently undergoing.

Dr. Anita Kumari, Associate Professor, Department of Chemical Engineering, Thadomal Shahani Engineering College, Bandra, Mumbai, India. She has obtained her **PhD from IIT, Delhi**. She has over 15 years of teaching experience. Her fields of specialization are Thermodynamics, Environmental Engineering, Heat and Mass Transfer.